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(FILE 'HOME' ENTERED AT 14:12:17 ON 30 SEP 2000)
      FILE 'CAPLUS' ENTERED AT 14:12:31 ON 30 SEP 2000
=> s metallocene#
            9020 METALLOCENE#
L1
=> s l1 and metal halide adduct
        1078587 METAL
          108068 HALIDE
           58854 ADDUCT
               11 METAL HALIDE ADDUCT
                     (METAL(W)HALIDE(W)ADDUCT)
L2
                O L1 AND METAL HALIDE ADDUCT
=> s l1 and titanium tetrachloride adduct
          293717 TITANIUM
44379 TETRACHLORIDE
           58854 ADDUCT
               11 TITANIUM TETRACHLORIDE ADDUCT
                     (TITANIUM(W)TETRACHLORIDE(W)ADDUCT)
L3
                0 L1 AND TITANIUM TETRACHLORIDE ADDUCT
       titanium tetrachloride adduct
          293717 TITANIUM
           44379 TETRACHLORIDE
           58854 ADDUCT
L4
               11 TITANIUM TETRACHLORIDE ADDUCT
                     (TITANIUM(W)TETRACHLORIDE(W)ADDUCT)
=> d 1-11 bib abs
L4
      ANSWER 1 OF 11 CAPLUS COPYRIGHT 2000 ACS
AN
      2000:578349 CAPLUS
TT
      Ring opening metathesis polymerization of dicyclopentadiene catalyzed by
      titanium tetrachloride adduct complexes with
      nitrogen-containing ligands
     Dono, Keleypette; Huang, Jiling; Ma, Haiyan; Qian, Yanlong
Laboratory of Organometallic Chemistry, East China University of Science
and Technology, Shanghai, 200237, Peop. Rep. China
J. Appl. Polym. Sci. (2000), 77(14), 3247-3251
CODEN: JAPNAB; ISSN: 0021-8995
ΑU
CS
SO
PB
      John Wiley & Sons, Inc.
DT
      Journal
LA
      English
AB
      Ring opening metathesis polymn. (ROMP) of dicyclopentadiene (DCPD)
      catalyzed by titanium tetrachloride adduct
      complexes such as TiCl4.cntdot.2L [L = pyridine (1), 2-methylpyridine
(2),
      2,4,6-trimethylpyridine (3), 3-aminopyridine (4), 2-hydroxypyridine (5)] and CH3Li as cocatalyst was reported. The polymer was characterized by
IR
      and 1H-NMR methods. Five influencing factors were also discussed. The catalyst systems TiCl4.cntdot.2L/CH3Li (L = 2-methylpyridine,
      2,4,6-trimethylpyridine) appeared to be very active for the ROMP of DCPD.
RE.CNT
RE
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- (1) Boutarfa, D; J Mol Catal 1991, V69, P157 CAPLUS (2) Breslow, D; Prog Polymer Sci 1993, V18, P1141 CAPLUS (3) Cannizzo, L; Macromolecules 1988, V21, P1961 CAPLUS
- (6) Gilliom, L; J Am Chem Soc 1986, V108, P733 CAPLUS (9) Larroche, C; J Org Chem 1982, V47, P2019 CAPLUS
- ALL CITATIONS AVAILABLE IN THE RE FORMAT
- ANSWER 2 OF 11 CAPLUS COPYRIGHT 2000 ACS 1999:706977 CAPLUS L4
- AN
- DN 132:36120
- TI Ring opening metathesis polymerization of dicyclopentadiene catalyzed by titanium tetrachloride adduct complexes containing nitrogen or oxygen ligand
- ΑU
- Keleypette, Dono; Huang, Jiling; Ma, Haiyan; Qian, Yanlong Laboratory of Organometallic Chemistry, ECUST, Shanghai, 200237, Peop. CS Rep. China
- Huadong Ligong Daxue Xuebao (1999), 25(4), 427-430 CODEN: HLIXEV; ISSN: 1006-3080 Huadong Ligong Daxue Xuebao Bianjibu SO
- PB
- DT
- LA Chinese
- The ring opening metathesis polymn. of dicyclopentadiene (DCPD) catalyzed AB by TiCl4.cntdot.2[C6H8O], TiCl4.cntdot.2[C8H11N], TiCl4.cntdot.2[C5H5N], TiCl4.cntdot.2[O(CH2)4CH2], TiCl4.cntdot.2[C6H7N], TiCl4.cntdot.2[C5H5N], and TiCl4.cntdot.2[C4H8O2] (6) is reported. These catalysis appear to exhibit good catalytic activity in the polymn. reaction. After a detailed
 - investigation of some influencing parameters, the optimized reaction conditions were obtained.
- ANSWER 3 OF 11 CAPLUS COPYRIGHT 2000 ACS 1997:792646 CAPLUS L4
- AN
- 128:120183 DN
- TI Thermochemical parameters of formation of titanium tetrachloride adducts as a measure of donor-acceptor bond strength
- ΑU
- CS
- Sevast'yanova, T. N.; Suvorov, A. V. St. Petersburg State University, St. Petersburg, 199164, Russia Russ. J. Coord. Chem. (Transl. of Koord. Khim.) (1997), 23(11), 761-770 CODEN: RJCCEY; ISSN: 1070-3284 S0
- MAIK Nauka/Interperiodica Publishing PB
- DT Journal; General Review
- LA Enalish
- A review and discussion with 77 refs. The mol. complexes (adducts) of AB titanium tetrachloride with mono- and bidentate donors are considered, and
 - the enthalpies of their formation from the components dissolved in nonag. solvents are compared with those of the corresponding cryst. complexes formed from the gaseous components. The enthalpy of formation of the cryst. complexes is shown to be strongly influenced by the intermol. interactions in the condensed state. It is established exptl. that the transition of the titanium tetrachloride complexes with acetonitrile, pyridine, and 2,2'- bipyridine to vapor has a dissociative character. Pyridine and 2,2'-bipyridine firmly hold titanium tetrachloride in the crystal state. The thermal effect of the reactions in nonaq. solvents is suggested to serve as a measure of the donor-acceptor interaction.
- L4 ANSWER 4 OF 11 CAPLUS COPYRIGHT 2000 ACS
- 1994:31305 CAPLUS ΑN
- DN 120:31305
- Magnesium-ethoxide-based titanium catalysts for polymerization of TI propylene
- ΑU
- Gupta, V. K.; Satish, S.; Bhardwaj, I. S. Res. Cent., Indian Petrochem. Corp. Ltd., Vadodara, 391346, India J. Macromol. Sci., Pure Appl. Chem. (1994), A31(4), 451-63 CS SO
- CODEN: JSPCE6: ISSN: 1060-1325
- DT Journal
- LA English
- AB Supported titanium catalysts are prepd. by the reactions of magnesium ethoxide and excess titanium tetrachloride with or without an internal Lewis base, di-Bu phthalate. The catalysts are characterized by compositional anal. and BET surface area measurements. The performance

catalysts using triethylaluminum and dimethoxydiphenylsilane as the cocatalyst system are examd. for propylene polymn. in slurry reactions. The di-Bu phthalate and phenyl-substituted methoxysilane used as internal and external Lewis bases, resp., govern the activity and stereospecificity of the catalyst system. Such polymn. parameters as time, temp., and hydrogen concn. also influence the performance of the catalytic system in terms of yield, isotactic index, and melt flow index of the polypropylene. ANSWER 5 OF 11 CAPLUS COPYRIGHT 2000 ACS L4 1991:596789 CAPLUS AN 115:196789 DN Titanium tetrachloride adduct with TI p-nitrosodiphenylamine Kogan, L. M.; Kuz'min, S. V.; Krol, V. A.; Shul'diner, M. D. Vses. Nauchno-Issled. Inst. Sint. Kauch, USSR Koord. Khim. (1991), 17(7), 914-17 CODEN: KOKHDC; ISSN: 0132-344X ΑU CS **SO** DT LA Russian The insertion reaction of p-nitrosodiphenylamine (L) in a 1:1 ratio gave TiCl3[N+(O-)(Cl)C6H4-p-NHPh] (I). TiCl4.nL (n = 2,3) and TiCl4.Q (Q = AB Ph2NH, N,N-diethylnitrosoaniline) were prepd. Paramagnetic I was characterized by electronic, IR, ESR, and mass spectra. L4 ANSWER 6 OF 11 CAPLUS COPYRIGHT 2000 ACS 1987:178386 CAPLUS ΑN DΝ 106:178386 Titanium-glycol crosslinking agents for polygalactomannans TI Morgan, Michael E. IN Celanese Corp. , USA Eur. Pat. Appl., 17 pp. PA SO CODEN: EPXXDW DT Patent LA English FAN.CNT² PATENT NO. KIND DATE APPLICATION NO. DATE PΙ EP 205335 A2 19861217 EP 1986-304399 19860610 EP 205335 19870506 Α3 R: BE, CH, DE, FR, GB, IT, LI, NL, SE US 1985-743246 US 4605736 19860812 Α 19850611 US 4677201 US 1986-830823 19870630 19860219 PRAI US 1985-743246 US 1986-830823 19850611 19860219 Aq. polygalactomannans crosslinked by reaction products of Ticl4 with water-sol. alcs. or diols are useful in fracturing solns. for secondary oil recovery and in gel explosives. Adding 100 parts Ticl4 over 1 h to 400 parts HOCH2CH2OH, cooling to 40.degree., adding 124 parts 50% NaOH over 10 min, and heating 1 h at 60.degree. gave a crosslinking compn. A soln. (viscosity 30-35 cP) of 40 lb hydroxypropyl guar gum in 1000 gal 2% AB aq. KCl was adjusted to pH 4.2 with AcOH and stirred with 2.6 gal crosslinker soln. to give a crosslinked polygalactomannan soln. with viscosity (130.degree.) 268, 428, and 580 cP after 0, 30, and 60 min, L4 ANSWER 7 OF 11 CAPLUS COPYRIGHT 2000 ACS AN 1979:24159 CAPLUS DN 90:24159 TI Metal halide-amide reaction product IN Bulson, Walter T.; Christie, Peter A.; Jones, James R. PA Armstrong Cork Co., USA SO U.S., 6 pp. CODEN: USXXAM DT Patent English LA FAN.CNT 1

PATENT NO.

US 4115423

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DATE

19780919

APPLICATION NO.

US 1976-723266

DATE

19760914

Products useful in the modification of aminoplasts are manufd. with less pollutant emission by reaction of metal halides with amides contg. gtoreq.2 replaceable H atoms in water-immiscible inert liqs., hydrolyzing the reaction products, and partially neutralizing the aq. solns. with bicarbonates. Thus, adding 12.1 g TiCl4 over 10-20 min to 45.6 g urea and 46 g C6H6 stirred at .ltoreq.40.degree., stirring 10 min, adding 31.7 g H2O, and neutralizing the aq. layer with 10.6 g NaHCO3 to pH <1.5 gives an aq. soln. of the reaction product. Adding 100 parts this soln. to 180 parts soln. prepd. from melamine 22.5, urea 18.0, 37% HCHO 116.0, MeOH 14.3, and KNO3 9.1 parts gives a cured resin [25036-13-9] with low shrinkage. ANSWER 8 OF 11 CAPLUS COPYRIGHT 2000 ACS 1977:34867 CAPLUS L4 ΑN DN 86:34867 Effect of steric factors in ligands on the ionization of titanium ΤI tetrachloride-ester complexes Lysenko, Yu. A.; Khokhlova, L. I.; Vedmedskaya, A. N. Donetsk. Politekh. Inst., Donetsk, USSR Izv. Vyssh. Uchebn. Zaved., Khim. Khim. Tekhnol. (1976), 19(9), 1330-2 ΑU CS SO CODEN: IVUKAR DT Journal LA Russian AB Migration studies in elec. fields of ions formed in solns. of TiCl4 in HCO2Pr, EtCO2C9H19, and C7H15CO2Et, and data previously obtained in TiCl4 solns. in 11 other esters indicate that electron donor and geometric properties of the ligands affect the ionization of complexes of the form TiCl4.E and TiCl4.2E (E = ester). The ester complexes entered into the internal coordination sphere of the ions [TiCl3.E]+ and [TiCl3.3E]+ which were generated in trans-conformal form. With increasing length of the alkyl group, the equil. is displaced in the direction of forming [TiCl3.E]+ ions.L4 ANSWER 9 OF 11 CAPLUS COPYRIGHT 2000 ACS 1976:413124 CAPLUS AN DN 85:13124 Lewis base properties of platinum(0) complexes. III. Adducts between TI platinum and titanium tetrachloride ΑU Plummer, J. F.; Schram, E. P. Dep. Chem., Ohio State Univ., Columbus, Ohio, USA Inorg. Chem. (1975), 14(7), 1505-12 CODEN: INOCAJ CS **SO** DT Journal English LA AB The reactions of Pt(PPh3)x (x = 3,4) with Ticl4 gave (TiCl4)2Pt[TiCl4(PPh3)]3 (I). Thermolysis of I gave Pt[TiCl4(PPh3)]3 (II). The reaction of I with PPh3 or PMePh2 also gave II. Treatment of Ι with BCl3 gave TiCl4; subsequent thermolysis of the reaction residue results in the evolution of addnl. TiCl4, BCl3-PPh3, and Pt[Tic14(PPh3)]2. L4 ANSWER 10 OF 11 CAPLUS COPYRIGHT 2000 ACS ΑN 1971:405161 CAPLUS DN 75:5161 TI Ketene adducts with halides. I. Reaction of ketene with tin tetrachloride ΑU Pavlov, V. I.; Koshkina, L. P. CS USSR Tr. Tol'yattinsk. Politekh. Inst. (1969), No. 1, 94-6 From: Ref. Zh., Khim. 1970, Abstr. No. 9Zh503 50 DT Journal LA Russian Reaction of H2C:C:O (I) with SnCl4, BiBr3, and TiCl4 gave stable adducts. I adducts of SnCl4 and BiBr3 have fluxing properties for soldering Al AB

other metals; the adduct of TiCl4 is a catalyst for isoprene polymn.

with

L4 ANSWER 11 OF 11 CAPLUS COPYRIGHT 2000 ACS 1969:29566 CAPLUS AN 70:29566 DN Catalytic ethylene polymerization ΤI Kosaka, Yujiro; Ohara, Hyakumon; Shibata, Taizo; Fujita, Nobuhiro IN Toyo Soda Manufg. Co., Ltd. PΑ Japan., 4 pp. CODEN: JAXXAD SO DT Patent Japanese LA FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE В4 PΙ 19680701 JΡ JP 43015623 19640615 Polyethylene of d. 0.91-0.98 can be prepd. by using a complex catalyst consisting of an adduct of TiCl4 with a compd. of the general formula MSR (where M is Na or K and SR is a mercapto radical) (mole ratio AB 0.05 - 4.0:1a trialkyl Al compd., and a tetraalkoxy Ti compd. Thus, a mixt. of 6.0 millimoles n-BuSNa, 4.0 millimoles TiCl4, and 250 ml. n-C7H16 was stirred for 30 min. at 50.degree., then 18 millimoles iso-Bu3Al and 3.0 millimoles Ti(OBu)4 were added and dild. to 300 ml. while stirring with n-C7H16. C2H4 at 2.0 kg./cm.2 was polymd. during 3 hrs. at 60.degree., giving 100 g. polymer of d. 0.926 and intrinsic viscosity 11.9 dl./g. in Tetralin at

130.degree..